Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

4-Allyl-2-methoxyphenyl 3,4-dichlorobenzenesulfonate

Ya-Tuan Ma,[a] Zhao-Feng Gao,[b] Qi-Chao Liu,[b] Gang Jin[b] and Jin-Ming Gao[b]*

[a]College of Science and College of Life Sciences, Northwest A&F University, Yangling Shaanxi 712100, People's Republic of China, and [b]College of Science, Northwest A&F University, Yangling Shaanxi 712100, People's Republic of China
Correspondence e-mail: mnathantuan@yahoo.com

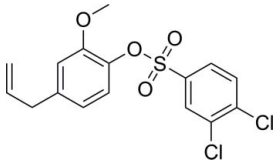
Received 11 October 2011; accepted 24 October 2011

Key indicators: single-crystal X-ray study; T = 298 K; mean σ(C–C) = 0.006 Å; R factor = 0.048; wR factor = 0.140; data-to-parameter ratio = 14.2.

The title compound, $C_{16}H_{14}Cl_2O_4S$, was obtained by the reaction of eugenol (4-allyl-2-methoxyphenol) and 3,4-dichlorobenzenesulfonyl chloride. The dihedral angle between the benzene rings in the molecule is 40.53 (4)°. No significantly short intermolecular contacts are observed in the crystal structure.

Related literature

For the synthesis of eugenol derivatives, see: Sadeghian *et al.* (2008). For a related structure, see: Ma *et al.* (2010).



Experimental

Crystal data

$C_{16}H_{14}Cl_2O_4S$ M_r = 373.23
Triclinic, $P\overline{1}$
a = 8.8694 (8) Å
b = 9.7501 (9) Å
c = 10.3796 (11) Å
α = 83.369 (2)°
β = 76.196 (1)°
γ = 80.038 (1)°
V = 855.95 (14) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.52 mm^{-1}
T = 298 K
0.45 × 0.40 × 0.30 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.801, T_{max} = 0.860
4290 measured reflections
2967 independent reflections
1762 reflections with $I > 2\sigma(I)$
R_{int} = 0.017

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.048
$wR(F^2)$ = 0.140
S = 1.02
2967 reflections
209 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.36 e Å^{-3}
$\Delta\rho_{min}$ = −0.33 e Å^{-3}

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

We acknowledge funding support from the National Natural Science Foundation of China (grant No. 30971882), Shaanxi Province Science and Technology (under contract No. 2011k02–07) and the Program of Northwest A&F University (No. Z111020908).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2390).

References

Bruker (2001). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Ma, Y.-T., Li, H.-Q., Shi, X.-W., Zhang, A.-L. & Gao, J.-M. (2010). *Acta Cryst.* E**66**, o2946.
Sadeghian, H., Seyedi, S. M., Saberi, M. R., Arghiani, Z. & Riazi, M. (2008). *Bioorg. Med. Chem.* **16**, 890–901.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

supplementary materials

Acta Cryst. (2011). E**67**, o3082 [**doi:10.1107/S1600536811044163**]

4-Allyl-2-methoxyphenyl 3,4-dichlorobenzenesulfonate

Y.-T. Ma, Z.-F. Gao, Q.-C. Liu, G. Jin and J.-M. Gao

Comment

In this paper, we present the structure of the title compound (Fig. 1), which was synthesized by the reaction of eugenol and 3,4-dichlorobenzenesulfonyl chloride (Sadeghian *et al.*, 2008). We previously reported a compound of this type (Ma *et al.*, 2010). In the molecular structure, the bond lengths and angles are normal and the dihedral angle between the aromatic rings is 40.53 (4)°. The crystal packing exhibits no significantly short intermolecular contacts.

Experimental

492 mg of eugenol (3 mmol), triethylamine (4 mmol), 3,4-dichlorobenzenesulfonyl chloride (3 mmol), and 40 ml of dichloromethane were mixed in a 100 ml flask. After 2 h under stirring at 278 K, the crude product was obtained. The crystals were obtained by recrystallization from methanol.

Refinement

The positions of all H atoms were fixed geometrically and C—H bond lengths fixed to 0.93 (aromatic CH), 0.96 (methyl CH_3) or 0.97 Å (methylene CH_2). Isotropic displacement parameters for H atoms were fixed to U_{iso}(H7x) = 1.5U_{eq}(C7) and U_{iso}(H) = 1.2U_{eq}(carrier C) for other H atoms.

Figures

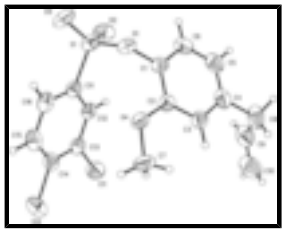


Fig. 1. The molecular structure of the title molecule. Displacement ellipsoids are drawn at the 30% probability level.

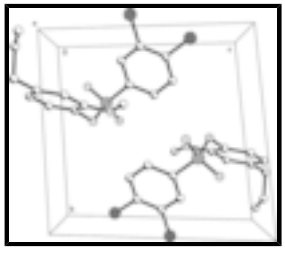


Fig. 2. Packing diagram.

4-Allyl-2-methoxyphenyl 3,4-dichlorobenzenesulfonate

Crystal data

$C_{16}H_{14}Cl_2O_4S$	$Z = 2$
$M_r = 373.23$	$F(000) = 384$
Triclinic, $P\overline{1}$	$D_x = 1.448$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 8.8694$ (8) Å	Cell parameters from 1336 reflections
$b = 9.7501$ (9) Å	$\theta = 2.8$–$23.7°$
$c = 10.3796$ (11) Å	$\mu = 0.52$ mm^{-1}
$\alpha = 83.369$ (2)°	$T = 298$ K
$\beta = 76.196$ (1)°	Prism, colourless
$\gamma = 80.038$ (1)°	0.45 × 0.40 × 0.30 mm
$V = 855.95$ (14) Å^3	

Data collection

Bruker SMART APEX CCD area-detector diffractometer	2967 independent reflections
Radiation source: fine-focus sealed tube	1762 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.017$
ω and φ scans	$\theta_{max} = 25.0°$, $\theta_{min} = 2.8°$
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)	$h = -10 \rightarrow 9$
$T_{min} = 0.801$, $T_{max} = 0.860$	$k = -11 \rightarrow 11$
4290 measured reflections	$l = -9 \rightarrow 12$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.048$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.140$	H-atom parameters constrained
$S = 1.02$	$w = 1/[\sigma^2(F_o^2) + (0.0575P)^2 + 0.4401P]$ where $P = (F_o^2 + 2F_c^2)/3$
2967 reflections	$(\Delta/\sigma)_{max} = 0.001$
209 parameters	$\Delta\rho_{max} = 0.36$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.33$ e Å^{-3}
0 constraints	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}

Cl1	0.82040 (13)	0.91886 (11)	1.44301 (11)	0.0879 (4)
Cl2	0.75166 (16)	1.04135 (13)	1.72255 (11)	0.0992 (4)
O1	0.5562 (2)	1.4568 (2)	1.2109 (2)	0.0595 (6)
O2	0.4219 (3)	1.2645 (3)	1.1872 (3)	0.0871 (9)
O3	0.2975 (3)	1.4507 (3)	1.3373 (3)	0.0915 (9)
O4	0.8069 (3)	1.4105 (3)	1.3153 (2)	0.0630 (6)
S1	0.43068 (11)	1.35611 (11)	1.28005 (11)	0.0703 (3)
C1	0.7014 (4)	1.3996 (3)	1.1322 (3)	0.0552 (9)
C2	0.8326 (4)	1.3787 (3)	1.1873 (3)	0.0548 (8)
C3	0.9776 (4)	1.3308 (4)	1.1077 (4)	0.0653 (10)
H3	1.0669	1.3146	1.1428	0.078*
C4	0.9901 (5)	1.3068 (4)	0.9758 (4)	0.0733 (11)
C5	0.8575 (6)	1.3299 (4)	0.9242 (4)	0.0807 (12)
H5	0.8659	1.3141	0.8359	0.097*
C6	0.7135 (5)	1.3759 (4)	1.0020 (4)	0.0716 (11)
H6	0.6244	1.3909	0.9669	0.086*
C7	0.9350 (5)	1.3767 (6)	1.3797 (4)	0.0961 (15)
H7A	0.9763	1.2793	1.3742	0.144*
H7B	0.8993	1.3964	1.4715	0.144*
H7C	1.0157	1.4315	1.3369	0.144*
C8	1.1522 (6)	1.2558 (5)	0.8904 (4)	0.0989 (16)
H8A	1.1523	1.2820	0.7973	0.119*
H8B	1.2304	1.3008	0.9136	0.119*
C9	1.1938 (7)	1.1033 (5)	0.9099 (4)	0.1003 (16)
H9	1.1251	1.0486	0.8940	0.120*
C10	1.3139 (8)	1.0416 (7)	0.9462 (5)	0.133 (2)
H10A	1.3855	1.0926	0.9632	0.159*
H10B	1.3318	0.9445	0.9563	0.159*
C11	0.5144 (4)	1.2605 (4)	1.4070 (3)	0.0578 (9)
C12	0.6161 (4)	1.1383 (4)	1.3789 (3)	0.0564 (9)
H12	0.6347	1.1026	1.2964	0.068*
C13	0.6895 (4)	1.0704 (4)	1.4771 (4)	0.0588 (9)
C14	0.6590 (4)	1.1249 (4)	1.6005 (4)	0.0651 (10)
C15	0.5542 (5)	1.2442 (4)	1.6267 (4)	0.0728 (11)
H15	0.5319	1.2779	1.7105	0.087*
C16	0.4817 (4)	1.3143 (4)	1.5303 (4)	0.0682 (10)
H16	0.4120	1.3965	1.5473	0.082*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0900 (8)	0.0784 (7)	0.0890 (8)	0.0078 (6)	−0.0208 (6)	−0.0102 (6)
Cl2	0.1245 (11)	0.1074 (9)	0.0685 (7)	−0.0214 (8)	−0.0319 (7)	0.0095 (6)
O1	0.0484 (14)	0.0575 (14)	0.0743 (16)	0.0003 (11)	−0.0208 (12)	−0.0098 (12)
O2	0.089 (2)	0.091 (2)	0.099 (2)	−0.0254 (16)	−0.0462 (17)	−0.0127 (16)
O3	0.0432 (15)	0.109 (2)	0.120 (2)	0.0058 (15)	−0.0224 (15)	−0.0165 (18)
O4	0.0466 (13)	0.0839 (17)	0.0603 (15)	−0.0047 (12)	−0.0146 (11)	−0.0154 (13)
S1	0.0509 (6)	0.0797 (7)	0.0867 (7)	−0.0086 (5)	−0.0258 (5)	−0.0125 (5)

C1	0.057 (2)	0.049 (2)	0.060 (2)	−0.0025 (16)	−0.0165 (17)	−0.0059 (16)
C2	0.054 (2)	0.049 (2)	0.060 (2)	−0.0053 (16)	−0.0118 (17)	−0.0059 (16)
C3	0.058 (2)	0.061 (2)	0.068 (2)	−0.0009 (18)	−0.0059 (19)	0.0014 (18)
C4	0.086 (3)	0.053 (2)	0.063 (3)	0.005 (2)	0.004 (2)	0.0011 (19)
C5	0.103 (4)	0.074 (3)	0.059 (2)	−0.001 (3)	−0.015 (2)	−0.008 (2)
C6	0.084 (3)	0.068 (3)	0.064 (2)	0.000 (2)	−0.026 (2)	−0.005 (2)
C7	0.057 (2)	0.158 (5)	0.082 (3)	−0.004 (3)	−0.034 (2)	−0.023 (3)
C8	0.101 (3)	0.076 (3)	0.086 (3)	0.011 (3)	0.023 (3)	0.002 (2)
C9	0.109 (4)	0.095 (4)	0.073 (3)	0.018 (3)	0.008 (3)	−0.017 (3)
C10	0.156 (6)	0.141 (5)	0.078 (3)	0.031 (5)	−0.014 (4)	−0.017 (3)
C11	0.048 (2)	0.062 (2)	0.066 (2)	−0.0153 (17)	−0.0104 (17)	−0.0087 (18)
C12	0.051 (2)	0.062 (2)	0.059 (2)	−0.0159 (18)	−0.0095 (17)	−0.0103 (17)
C13	0.054 (2)	0.057 (2)	0.065 (2)	−0.0165 (17)	−0.0068 (18)	−0.0049 (18)
C14	0.070 (2)	0.072 (3)	0.055 (2)	−0.026 (2)	−0.0103 (18)	0.0020 (19)
C15	0.081 (3)	0.076 (3)	0.059 (2)	−0.018 (2)	−0.002 (2)	−0.014 (2)
C16	0.062 (2)	0.066 (2)	0.071 (3)	−0.0100 (19)	0.001 (2)	−0.017 (2)

Geometric parameters (Å, °)

Cl1—C13	1.729 (4)	C7—H7A	0.9600
Cl2—C14	1.724 (4)	C7—H7B	0.9600
O1—C1	1.411 (4)	C7—H7C	0.9600
O1—S1	1.598 (2)	C8—C9	1.471 (6)
O2—S1	1.411 (3)	C8—H8A	0.9700
O3—S1	1.419 (3)	C8—H8B	0.9700
O4—C2	1.356 (4)	C9—C10	1.247 (7)
O4—C7	1.424 (4)	C9—H9	0.9300
S1—C11	1.761 (4)	C10—H10A	0.9300
C1—C6	1.372 (5)	C10—H10B	0.9300
C1—C2	1.390 (4)	C11—C12	1.379 (4)
C2—C3	1.389 (5)	C11—C16	1.387 (5)
C3—C4	1.391 (5)	C12—C13	1.383 (5)
C3—H3	0.9300	C12—H12	0.9300
C4—C5	1.378 (6)	C13—C14	1.391 (5)
C4—C8	1.531 (5)	C14—C15	1.367 (5)
C5—C6	1.371 (6)	C15—C16	1.371 (5)
C5—H5	0.9300	C15—H15	0.9300
C6—H6	0.9300	C16—H16	0.9300
C1—O1—S1	119.2 (2)	H7B—C7—H7C	109.5
C2—O4—C7	117.7 (3)	C9—C8—C4	111.4 (4)
O2—S1—O3	121.23 (17)	C9—C8—H8A	109.4
O2—S1—O1	109.01 (16)	C4—C8—H8A	109.4
O3—S1—O1	102.96 (16)	C9—C8—H8B	109.4
O2—S1—C11	109.38 (17)	C4—C8—H8B	109.4
O3—S1—C11	109.49 (18)	H8A—C8—H8B	108.0
O1—S1—C11	103.13 (14)	C10—C9—C8	125.2 (6)
C6—C1—C2	121.5 (3)	C10—C9—H9	117.4
C6—C1—O1	120.5 (3)	C8—C9—H9	117.4
C2—C1—O1	117.9 (3)	C9—C10—H10A	120.0

O4—C2—C3	125.6 (3)	C9—C10—H10B	120.0
O4—C2—C1	116.1 (3)	H10A—C10—H10B	120.0
C3—C2—C1	118.3 (3)	C12—C11—C16	122.1 (3)
C2—C3—C4	120.3 (4)	C12—C11—S1	119.1 (3)
C2—C3—H3	119.8	C16—C11—S1	118.7 (3)
C4—C3—H3	119.8	C11—C12—C13	118.1 (3)
C5—C4—C3	119.7 (4)	C11—C12—H12	121.0
C5—C4—C8	121.3 (4)	C13—C12—H12	121.0
C3—C4—C8	118.9 (4)	C12—C13—C14	120.1 (3)
C6—C5—C4	120.5 (4)	C12—C13—Cl1	118.9 (3)
C6—C5—H5	119.7	C14—C13—Cl1	121.0 (3)
C4—C5—H5	119.7	C15—C14—C13	120.5 (3)
C5—C6—C1	119.7 (4)	C15—C14—Cl2	119.3 (3)
C5—C6—H6	120.2	C13—C14—Cl2	120.2 (3)
C1—C6—H6	120.2	C14—C15—C16	120.4 (4)
O4—C7—H7A	109.5	C14—C15—H15	119.8
O4—C7—H7B	109.5	C16—C15—H15	119.8
H7A—C7—H7B	109.5	C15—C16—C11	118.8 (4)
O4—C7—H7C	109.5	C15—C16—H16	120.6
H7A—C7—H7C	109.5	C11—C16—H16	120.6
C1—O1—S1—O2	−44.3 (3)	C3—C4—C8—C9	83.8 (5)
C1—O1—S1—O3	−174.3 (2)	C4—C8—C9—C10	−123.3 (6)
C1—O1—S1—C11	71.8 (3)	O2—S1—C11—C12	27.3 (3)
S1—O1—C1—C6	82.4 (4)	O3—S1—C11—C12	162.3 (3)
S1—O1—C1—C2	−102.5 (3)	O1—S1—C11—C12	−88.6 (3)
C7—O4—C2—C3	−8.0 (5)	O2—S1—C11—C16	−155.9 (3)
C7—O4—C2—C1	173.5 (3)	O3—S1—C11—C16	−20.9 (3)
C6—C1—C2—O4	177.7 (3)	O1—S1—C11—C16	88.2 (3)
O1—C1—C2—O4	2.6 (5)	C16—C11—C12—C13	−1.5 (5)
C6—C1—C2—C3	−1.0 (5)	S1—C11—C12—C13	175.2 (2)
O1—C1—C2—C3	−176.0 (3)	C11—C12—C13—C14	0.6 (5)
O4—C2—C3—C4	−177.4 (3)	C11—C12—C13—Cl1	−178.8 (3)
C1—C2—C3—C4	1.0 (5)	C12—C13—C14—C15	1.2 (5)
C2—C3—C4—C5	−0.5 (6)	Cl1—C13—C14—C15	−179.4 (3)
C2—C3—C4—C8	179.2 (3)	C12—C13—C14—Cl2	−179.4 (3)
C3—C4—C5—C6	−0.2 (6)	Cl1—C13—C14—Cl2	−0.1 (4)
C8—C4—C5—C6	−179.9 (4)	C13—C14—C15—C16	−2.2 (6)
C4—C5—C6—C1	0.3 (6)	Cl2—C14—C15—C16	178.5 (3)
C2—C1—C6—C5	0.3 (6)	C14—C15—C16—C11	1.3 (6)
O1—C1—C6—C5	175.2 (3)	C12—C11—C16—C15	0.6 (5)
C5—C4—C8—C9	−96.5 (5)	S1—C11—C16—C15	−176.2 (3)

Fig. 1

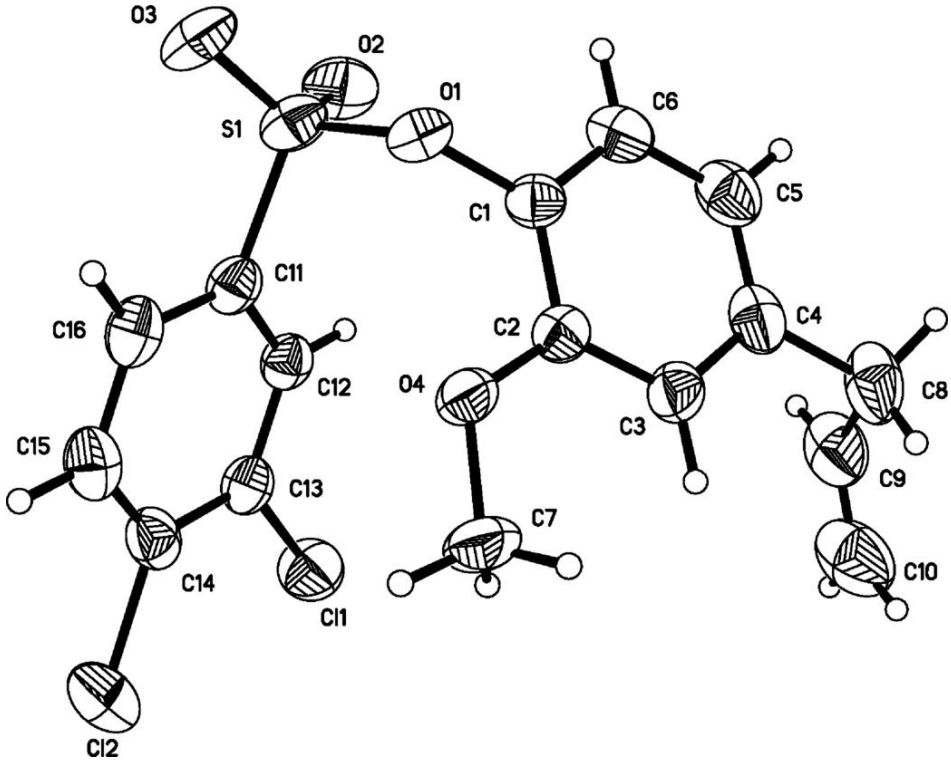

O3
O2
S1
O1
C6
C5
C1
C11
C16
C2
C4
C12
O4
C8
C3
C15
C13
C9
C7
C14
C10
Cl1
Cl2

Fig. 2